¹H-N.M.R. ANALYSIS OF TYPE-2 CHAIN LACTO-GANGLIOSIDES. CONFIRMATION OF STRUCTURE OF A NOVEL CANCER-ASSOCIATED FUCOGANGLIOSIDE, α-NeuAc-(2 \rightarrow 6)-β-D-Galp-(1 \rightarrow 4)-β-D-GlcpNAc-(1 \rightarrow 3)-β-D-Glcp-(1 \rightarrow 4)-[α-L-Fucp-(1 \rightarrow 3)]-β-D-GlcpNAc-(1 \rightarrow 3)-β-D-Galp-(1 \rightarrow 4)-β-D-Glcp-(1 \rightarrow 1)-Cer (VI⁶NeuAcIII³FucnLc₆Cer)*,†

STEVEN B. LEVERY, EDWARD NUDELMAN, REIJI KANNAGI, FRANK W. SYMINGTON, NIELS H. ANDERSEN. HENRIK CLAUSEN. MICHAEL BALDWIN, AND SEN-ITIROH HAKOMORI

Program of Biochemical Oncology/Membrane Research, Fred Hutchinson Cancer Research Center, and Departments of Pathobiology, Microbiology & Immunology, and Chemistry, University of Washington, Seattle, Washington 98104 (U.S.A.)

(Received August 26th, 1986; accepted for publication, December 1st, 1986)

ABSTRACT

Neolacto-glycosphingolipids, substituted with α-NeuAc- $(2\rightarrow 3)$ - and $-(2\rightarrow 6)$ -linked p-Galp residues were analyzed by one- and two-dimensional ¹H-n.m.r. spectroscopy at 500 MHz in 49:1 (v/v) di(²H₃)methyl sulfoxide-deuterium oxide solution. For the simplest structures analyzed, nLc₄Cer, IV³NeuAcnLc₄Cer, and IV⁶NeuAcnLc₄Cer, sialosylation-induced changes in shifts of terminal and subterminal core residues were interpretable in terms of existing conformational models. Chemical shifts for H-3e and H-3a of NeuAc characteristic for the type of linkage, were also determined. In addition, regularly reproducible shifts were seen for H-1 and other resonances of terminal and subterminal core residues of all structures tested. Chemical-shift correlations proved to be useful in elucidating the structure of a unique ganglioside bearing an internal β-D-Galp- $(1\rightarrow 4)$ -[α-L-Fucp- $(1\rightarrow 3)$]-β-D-GlcpNAc- $(1\rightarrow 3)$ residue ("X-trisaccharide") with an α-NeuAc- $(2\rightarrow 6)$ -substituted terminal group.

INTRODUCTION

Neolacto-gangliosides having an α -NeuAc-(2 \rightarrow 6)- β -D-Galp terminal group, a relatively recent addition to the list of known glycosphingolipid structures, appear to have a rather limited distribution in normal human tissues. α -NeuAc-(2 \rightarrow 6)-lacto-N-neo-tetraosyl- and -nor-hexaosylceramides were characterized as minor components of human erythrocytes¹, and subsequently the former was found to be present in moderate quantities in human meconium². In connection with the

^{*}Dedicated to Professor Walter T. J. Morgan.

[†]This investigation was supported by grants from the National Institutes of Health, CA-19 224 and CA-42 505. S. H. is a recipient of an Outstanding Investigator Award from the National Cancer Institute (CA-42 505).

present study, significant quantities of these compounds, along with α -NeuAc- $(2\rightarrow 3)$ -substituted analogs, were found in human granulocytes³, in agreement with the results of Fukuda et al.⁴. However, the largest accumulation so far of α -NeuAc-(2→6)-substituted *neolacto*-gangliosides have been found in human colonic and liver adenocarcinoma^{5,6}. In addition to large quantities of α -NeuAc-(2 \rightarrow 6)-lactoneo-tetraosylceramide (IV6NeuAcnLc₄Cer) (3)*, two novel gangliosides, VI6NeuAcIII3FucnLc₆Cer (10) and VI3NeuAcV3III3Fuc₂nLc₆Cer (11), were found and characterized^{5,6,8}. Since these compounds were virtually absent in ganglioside fractions of normal colonic mucosa and liver tissue, they may be tumor-associated antigens. To facilitate the analysis of such compounds of interest, we have adopted the routine use of ¹H-n.m.r. spectroscopy to obtain structural information complementary to that obtained by other methods, i.e., methylation analysis by m.s. and g.l.c.-m.s., t.l.c., enzymic and chemical degradation, and the more recent application of monoclonal antibody techniques. In the past few years, ¹H-n.m.r. spectroscopy has been of increasing importance in the elucidation of glycosphingolipid structures⁹⁻¹². In particular, the use of two-dimensional methods^{13,14} such as J-resolved^{15,16}, spin-echo correlated^{12,17}, and n.O.e.¹² spectroscopy have led to increased information. However, as has been pointed out¹⁵, the chief limitations will be the time and amount of material available. It is apparent that many glycolipids, particularly those found in tumors, will be available in very limited quantities. Thus, it is impractical to expect to apply, in every case, two-dimensional n.m.r. techniques or to suggest establishing all connectivities between residues by sequential application of spin-decoupling difference spectroscopy^{18,19} (s.d.d.s.) and n.O.e. experiments⁹. For this reason, it is anticipated that for some time to come, structural assignment will still depend heavily on correlative methods, e.g., identifying resonances characteristic of particular structural features (linkage position, and anomeric configuration), and recognition of glycosidation-induced chemical shift changes of readily assignable resonances, provided that the regularity of such effects has been established with some degree of certainty. This appears to be particularly applicable to glycolipids of the Type 2-lacto series, which have a regular repeating backbone structure that is perturbed only within a limited domain by the addition of side-chains and terminal groups.

The complete assignment of resonances of asialyl- and monosialyl-ganglioseries glycolipids has been published¹². However, with the exception¹¹ of IV³NeuAcnLc₄Cer (3), no ¹H-n.m.r. data are currently available for *neolacto-gangliosides*. The purpose of this communication is to: (a) present data on resonances characteristic for this series to aid in identification of the particular substances studied; (b) identify some of the general trends associated with NeuAc

^{*}Shorthand designations for glycolipids are used according to the recommendations of the IUPAC-IUB Commission on Biochemical Nomenclature⁷; however, the suffix "ose" is omitted in the text, while "Cer" is omitted as well in tables. The prefix α - is assumed for NeuAc and L Fuc residues and has also been omitted. The nonstandard designation "iso-nLc₈" is used in this paper for IV⁶[β -D-Galp-(1 \rightarrow 4)- β -D-GlcpNAc-(1 \rightarrow]nLc₆, to distinguish it from the unbranched structure nLc₈.

Glycolipid nomenclature	Abbreviated carbohydrate nomenclature*
1 nLc ₄ Cer	BGal4BGlcNAc3BGal4BGlc1Cer
2 IV3NeuAcnLc ₄ Cer	aNeuAc3βGal4βGlcNAc3βGal4βGlc1Cer
3 IV6NeuAcnLc2Cer	aNeuAc6BGal4BGlcNAc3BGal4BGlc1Cer
4 nLc,Cer	BGal4BGlcNAc3BGal4BGlcNAc3BGal4BGlc1Cer
5 VI³NeuAcnLc,Cer	aNeuAc3BGal4BGlcNAc3BGal4BGlcNAc3BGal4BGlcCer
6 VI ⁶ NeuAcnLc ₆ Cer	aNeuAc5BGal4BGlcNAc3BGal4BGlcNAc3BGal4BGlc1Cer
7 iso-nLc _s Cer	BGal4BGicNAc3(BGal4BGicNAc6)BGal4BGicNAc3BGal4BGIc1Cer
8 VI ³ NeuAc-iso-nLc ₈ Cer	aNeuAc3BGal4BGicNAc3(BGal4BGicNAc6)BGal4BGicNAc3BGal4BGic1Cet
9 VI3NeuAcVIII3NeuAc-iso-nLc ₈ Cer	aNeuAc3BGal4BGlcNAc3(aNeuAc3BGal4BGlcNAc6)BGal4BGlcNAc3BGal4BGlc1Cer
10 VI ⁶ NeuAcIII ³ FucnLc ₆ Cer	aNeuAc6BGal4BGlcNAc3BGal4(aFuc3)BGlcNAc3BGal4BGlc1Cer
11 VI ³ NeuAcIII ³ V ³ Fuc,nLc,Cer	lphaNeuAc3 eta Gal4($lpha$ Fuc3) eta GlcNAc3 eta Gal4($lpha$ Fuc3) eta GlcNAc3 eta Gal4 eta Glc1Cer
12 VI3NeuAcIII3FucnLc,Cer	αNeuAc3BGa14BGlcNAc3BGa14(αFuc3)BGlcNAc3BGa14BGlc1Cer

glucose, the L configuration for fucose, and the linkage at C-1 (C-2 for NeuAc) are assumed; the configuration of the anomeric centre (α or β) is placed before the name of the sugar, as in the systematic nomenclature of carbohydrates. Scheme 1. Structures of glycolipids. In the abbreviated nomenclature (*), the pyranose form, the D configuration for the hexoses and 2-acetamido-2-deoxy-

substitution on the *neolacto* series; and (c) use this information to confirm the structure of a fucoganglioside, isolated from adenocarcinoma, with a minimum amount of time and material.

EXPERIMENTAL

 1 H-N.m.r. spectroscopy. — Gangliosides were first converted to the Na⁺ form by passing through ion exchange resin AG-50W-X8 (Bio-Rad) Na⁺ form, in methanol–H₂O. Glycolipid samples (100–300 μg) were then deuterium exchanged by repeated addition and evaporation of 1:1 CDCl₃-CD₃OD and dissolved in (CD₃)₂SO containing 2% D₂O (ref. 18) and 1% tetramethylsilane as chemical shift references. For very small samples (~100 μg), particularly those containing an α-NeuAc-(2→6)-linked group, it was desirable to perform a final lyophilization from a solution in (CD₃)₂SO-D₂O, and to dissolve the sample in the highest quality (CD₃)SO (99.96 atom %) because, in these compounds, the H-3e resonance can be obscured by satellite peaks from a large residual Me₂SO peak.

500-MHz 1D 1 H-N.m.r. spectra were obtained at 308 and 328 \pm 2K with a Bruker (F.R.G.) WM-500 spectrometer–Aspect 2000 data system using quadrature detection, a spectral width of 5000 Hz over 16K data points, and a 2-s relaxation delay. The resolution of spectra was enhanced by application of a Lorentzian-to-Gaussian function²⁰, included in the Bruker software package, to the accumulated free-induction decays before Fourier transformation. The accuracy of the chemical shift measurements was \pm 0.001 p.p.m. Coupled resonances were determined conventionally by continuous irradiation of the appropriate resonance during acquisition.

Two-dimensional, phase sensitive COSY spectra²¹ were obtained at 308 \pm 2K with a Bruker AM-300 spectrometer-Aspect 3000 data system under conditions previously described²². Additional COSY spectra in the magnitude mode were collected with a Bruker WM-500-Aspect 2000 instrument without quadrature in t_1 using established acquisition parameters.

Thin-layer chromatography. — Preparative and analytical t.l.c. was performed on Merck (Darmstadt, F.R.G.). Silica gel 60 HPTLC plates developed in the following solvents: (A) 50:40:11 (v/v) chloroform-methanol-water containing 0.05% CaCl₂; (B) 7:3:1 (v/v) propanol-water-conc. NH₃. Bands scraped from plates for preparative purposes were extracted with 11:5:4 (v/v) 2-propanol-hexane-water (upper phase discarded). For h.p.t.l.c. immunostaining, it was necessary to use a plate having a silica layer high in binder, such as that provided by J. T. Baker, which will not wash off in aqueous solutions.

Other analytical procedures. — Sphingosine bases were analyzed as long chain aldehydes according to the method of Sweeley and Moscatelli²³ after methanolysis with the modified reagent as described by Gaver and Sweeley²⁴. G.l.c. was performed with a Hewlett-Packard 5890A using a 30M SPB-5 bonded-phase fused silica capillary column (temperature program, 140–180° at 2°/min). Identities of aldehyde peaks were verified (vs. synthetic standards) with a Finnigan 3300/6110 GC-MS/DS (DB-5 column, spitless injection) instrument.

Fatty acids were determined as methyl esters liberated by methanolysis (0.5m HCl, 80°, 24 h) of glycolipids. G.l.c. was performed with a DB-5 (similar to SPB-5) column (splitless injection, temperature program, 150–290° at 4°/min) with chemical-ionization detection (methane, 300 μ m)²⁵.

Reagents. — $Di(^2H_3)$ methyl sulfoxide (99.9 atom %), (2H)chloroform (99.9 atom %), and (2H_4)methanol (99.5 atom %) were purchased from Stohler Isotope (Waltham, MA) or Aldrich (Milwauke, WI). $Di(^2H_3)$ methyl sulfoxide (99.96 atom %) and D_2O (99.96 atom %) were purchased from Aldrich.

Glycosphingolipids. — IV³NeuAcnLc₄Cer (2), IV⁶NeuAcnLc₄Cer (3), and VI³NeuAcnLc₆Cer (5) were obtained from human neutrophil membranes according to methods previously described²⁶. Further details concerning their characterization have been published elsewhere³. Compounds 2 and 3 were also obtained from human erythrocyte membranes²⁷⁻³⁰. nLc₄Cer (1) was obtained by treatment of 2 from human erythrocytes²⁷ with acetic acid in 20:20:1 (v/v) chloroform-methanol-water at 100° in a sealed test tube. nLc₆Cer (4) was obtained from human adult type A and O erythrocyte membranes³¹. VI⁶NeuAcnLc₆Cer (6) was obtained from human liver adenocarcinomata by methods previously described^{5,6}. This compound has the some carbohydrate structure VI6NeuAcnLc₆Cer (6) obtained from human erythrocytes¹. VI3NeuAc-iso-nLc₈Cer (8) was obtained from human type O and A erythrocytes¹; VI³NeuAcVIII³NeuAciso-nLc_sCer (9) was obtained from human type O erythrocytes as described by Kundu et al.³². iso-nLc₈Cer (7) was obtained by acetic acid treatment of 9; its Hn.m.r. spectrum was essentially as described by Dabrowski et al.³³, but with shift differences due to differences in probe temperature. Structures of all glycolipids were confirmed by methylation analysis, including linkage analysis by g.l.c.-m.s. of reduced and acetylated hydrolyzates as described^{34–38}.

A preliminary account of the isolation and characterization of a fraction of VI⁶NeuAcIII³FucnLc₆Cer (**10**) has been published⁵. Subsequent to preliminary fractionation by l.c., the isolation of this glycolipid essentially involved preparative h.p.t.l.c. using solvent A (see Fig. 1A), followed by separation of the major band, migrating just below IV⁶NeuAcnLc₆Cer (**6**), into two components by preparative h.p.t.l.c. in solvent B (Fig. 1B). The faster migrating band in solvent B, reactive with MAb FH6, was shown to have the structure, VI³NeuAcV³FucIII³FucnLc₆Cer (**11**, "6B", ref. 8). To the slower component (Fraction "6C, upper"), reactive with MAb 1B9, was assigned⁵ structure **10**. The band migrating just below Fraction 6B–6C upper in solvent A (Fig. 1A) was similarly separated into two components, by use of solvent B, the faster one reactive with FH6 (Fraction "7B"), the slower one with 1B9 (Fraction "6C, lower", see Fig. 1B). Analysis by ¹H-n.m.r. showed that Fraction 7B had carbohydrate resonances essentially identical with those found previously²² for Fraction 6B. The complete analysis of the Fraction 6C components, as described below, is discussed in detail in the following sections.

Antibodies. — The production and determination of specificity of monoclonal antibodies (Mab) IB9 recognizing the terminal structure α -NeuAc-(2 \rightarrow 6)-D-Gal,

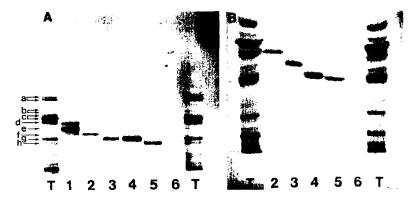


Fig. 1. H.p.t.l.c. of *lacto*-series gangliosides from adenocarcinoma FT75-620: (A) solvent A; (B) solvent B; (T) total monosialoganglioside fraction. Bands marked: (a) GM₃; (b) IV³NeuAcnLc₄Cer (2);(c) IV⁶NeuAcnLc₄Cer (3); (d–e) VI³NeuAcnLc₆Cer (5); (f) upper band of VI⁶NeuAcnLc₆Cer (6); (g) 6B + 6C upper; and (h) 7B + 6C lower. Lanes marked: (1) VI³NeuAcnLc₆Cer (5): (2) VI⁶NeuAcnLc₆Cer (6); (3) 6B; (4) 6C upper; (5) 6C lower; and (6) 7B.

and FH6 recognizing the structure, α -NeuAc- $(2\rightarrow 3)$ - β -D-Galp- $(1\rightarrow 4)$ - $[\alpha$ -L-Fucp- $(1\rightarrow 3)]$ - β -D-GlcpNAc- $(1\rightarrow 3)$ - β -D-Galp- $(1\rightarrow 4)$ - $[\alpha$ -L-Fucp- $(1\rightarrow 3)]$ -D-GlcNAc, have been described previously^{5.6.8}. The reactivity of glycolipids was determined by h.p.t.l.c.-immunostaining using the method of Magnani *et al.*³⁹, as modified by Kannagi *et al.*⁴⁰

RESULTS AND DISCUSSION

The anomeric proton chemical shifts and ${}^3J_{1,2}$ coupling constants for compounds 1–9 are summarized in Table I. Additional ring proton assignments made from two-dimensional COSY spectra are given for compounds 1–3. The phase-sensitive COSY spectrum of IV⁶NeuAcnLc₄Cer (3) is reproduced in Fig. 2 to illustrate the connectivities from which these assignments were extracted.

Results similar to those given for unsubstituted core structures 1, 4, and 7, differing by temperature-dependent shift effects, have been previously obtained^{9,33}. Results at a different temperature for anomeric protons of 2 have also been published¹¹. N-Acetylneuraminic acid H-3e and H-3e resonances, recognizable by their characteristic multiplicities and easily verified mutual coupling, are listed in Table II along with NeuAc and GlcNAc N-acetyl methyl resonances, and NeuAc H-4 resonances found in COSY spectra of 2 and 3. Values found for α -(2 \rightarrow 3)-linked NeuAc are similar to those reported for α -NeuAc-(2 \rightarrow 3)- β -D-Galp-(1 \rightarrow 4)- β -D-Glcp-(1 \rightarrow 1)-Cer (GM₃) obtained under similar conditions¹².

Unbranched analogs 1-6. — The values found for anomeric protons (Table I) show good numerical agreement for substitution on nLc_4 and nLc_6 cores. Likewise, the results for H-3e, H-3a, and N-acetyl resonances, shown in Table II, are in excellent agreement. Thus, these resonances would appear to be suitable in this solvent system as "structural reporter groups"⁴¹. Interestingly, the shift changes

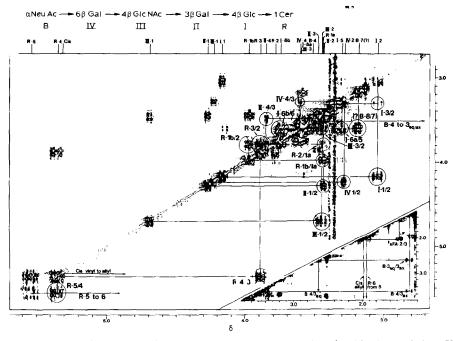


Fig. 2. Downfield region of two-dimensional, phase-sensitive $^1\text{H-COSY}$ at 300 MHz of $IV^6\text{NeuAcnLc}_4\text{Cer}$ (3) from O erythrocytes. Conc. $\sim\!2\text{mM}$ in di($^2\text{H}_3$)methyl sulfoxide-2% $D_2\text{O}$. Inset: section of lower frequency region. Arabic numerals refer to ring protons of residues designated by Roman numerals or capital letters in the corresponding structure. R refers to protons of sphingosine backbone only. *Cis*-vinyl and -allyl refer to protons of unsaturated fatty acids. nFA refers to protons of nonhydroxylated fatty acids.

seen for these compounds in dimethyl sulfoxide (Table III) are somewhat different from those reported for $^1\text{H-n.m.r.}$ of glycopeptides and oligosaccharides containing similar structures taken 41,42 in D_2O . The effects on the terminal Gal H-1 signal are smaller and opposite in sign, whereas the effects on the GlcNAc H-1 signal are of somewhat larger magnitude. Also, although the NAc signals of NeuAc occur at about the same location for α -(2 \rightarrow 3) or α -(2 \rightarrow 6) linkages for solutions in D_2O , they are sufficiently separated (by 0.010 \pm 0.001 p.p.m.) for solutions in di(2H_3)methyl sulfoxide to provide another basis for differentiation*.

Although less reliable as a structural marker, a characteristic feature of the α -NeuAc-(2 \rightarrow 6) linkage is the appearance of the subterminal Gal H-1 resonance as

*Since the results reported by Sabesan and Paulson⁴² for the structures, α -NeuAc-(2 \rightarrow 3 or 6)- β -D-Galp-(1 \rightarrow 4)- β -D-GlcpNAc-(1 \rightarrow 3)- β -D-Galp-(1 \rightarrow 4)-D-Glc, in D₂O are in essential agreement with those compiled by Vliegenthart *et al.*⁴¹ for both branched and unbranched structures, α -NeuAc-(2 \rightarrow 3 or 6)- β -D-Galp-(1 \rightarrow 4)- β -D-GlcpNAc-(1 \rightarrow 2 or 4)- α -D-Manp-(1 \rightarrow 3 or 6)- β -D-Manp-(1 \rightarrow , it would appear that the differences with our results reflect differences in solvent characteristics (*e.g.*, solvation), or solvent-induced conformational changes, or both. It should be noted, however, that in the case of substitution with an α -(2 \rightarrow 3)-linked NeuAc group, small but significant differences in shift changes were reported for β -D-GlcNAc H-1, dependent on which branch of the core structure is involved⁴¹.

 $^{1}\text{H.}\text{CHEMICAL SHIFTS }(\delta)^{a}\text{ and }\vartheta_{1,2}\text{ coupling constants }(Hz)^{b}\text{ for core sugars of Glycosphingolipids}^{r}$

		(A) Unbranched glycosphingolipids (groups or residues)	ycosphingolipids (groups or resid	(ues)			
Compound	Proton	α-NeuAc	β-D-Galp-(1→4)-β-D-GlcpNAc-(1	+3)-β-D-Galp-(1→4)	1-B-D-GlcpNAc-(1-	→3)-β-D-Galp-(1→4,	β-0-Gaip-(1→4)-β-D-GiqpNAc-(1→3)-β-0-Gaip-(1→4)-β-0-GicpNAc-(1→3)-β-0-Gaip-(1→4)-β-0-Gicp-(2→1)-Cer
nLc ₄ (1)	H-1 H-2 H-3	None	4.214 (6.7) 3.301 3.291 3.635	4.214 (6.7) 4.664 (8.5) 3.301 3.444 3.291 3.556 3.635			4.267 (7.3) 3.435 3.447 3.845	4.172 (7.9) ^d 3.044 3.314
IV ³ NeuAcnLc ₄ (2)	H-1 H-2 H-3	-(2→3)	4.201 (7.8) 3.291 3.964 3.700	4.642 (7.8) 3.449 3.543			4.262 (7.3) 3.430 3.452 3.849	4.166 (7.8)° 3.047 3.319
IV [©] NeuAcnLc ₄ (3)	H-1 H-2 H-3	-(2→6)	4.228 (7.9) 3.284 3.284 3.614	4.228 (7.9) 4.687 (8.5) 3.284 3.449 3.284 3.606 3.614			4.265 (7.9) 3.433 3.479 3.850	4.166 (7.9) 3.046 3.316
$\mathrm{nLc}_{6}^{f}\left(4 ight)$		None	4.208 (6.1)	4.208 (6.1) 4.652 (8.5)	4.260 (7.3) 4.652 (8.5)	4.652 (8.5)	4.260 (7.3)	4.260 (7.3) 4.169 (7.9)s
$VI^3NeuAcnLc_b^f(5)$		-(2→3)	4.200 (7.9)	4.200 (7.9) 4.636 (7.9)	4.262 (7.3)	4.262 (7.3) 4.652 (7.9)	4.263 (7.3)	4.168 (7.9)
VI ⁶ NeuAcnLc ₆ ^f (6)		.(2->6)	4.223 (7.3)	4.223 (7.3) 4.691 (8.5)	4.264 (7.3) 4.656 (8.5)	4.656 (8.5)	4.262 (7.3)	4.262 (7.3) 4.164 (7.9)

	(B) Branc	Branched glycosphingolipids (groups or residues) $^{\rm h}$	ıgolipids (grou _l	ps or residues)	ų				
Compound	α -Neu Ac^c	α-NeuAc ⁶ β-D-Galp-(1	4	4)-β-p-GlcpNAc-(1		(9)	į		:
	α-NeuAc³	35	β-D-Galp-(1		4)-β-υ-GlcpNAc-(1	β-d-Galp-(1→4. 3)	J-β-D-GicpNAc-(I→	3)-β-D-Cialp-(1→4,	B-d-Calp-(1→4)-B-D-CricpNAc-(1→3)-B-D-Calp-(1→4)-B-D-Cricp-(1→1)-Cer
$iso-nLc_g'(7)$	None ⁶ None ³	4.215 (7.3)	4.215 (7.3) 4.215 (7.3) 4.409 (7.9) 4.650 (8.5)	4.409 (7.9)	4.650 (8.5)	4.297 (7.9)	4.297 (7.9) 4.647 (8.5)	4.262 (7.3)	4.262 (7.3) 4.171 (7.9)*
VI ³ NeuAc-	None	4.214 (6.7)	4.198 (7.9)	4.198 (7.9) 4.416 (8.5) 4.662 (7.3)	4.662 (7.3)	4.301 (7.9)	4.301 (7.9) 4.646 (7.3)	4.264 (7.3)	4.264 (7.3) 4.169 (7.3)
$iso-nLc_g'(8)$	$(2\rightarrow3)^3$								
VI ³ , VIII ³ .	$(2\rightarrow 3)^6$	4.200 (7.9)	4.200 (7.9) 4.200 (7.9) 4.388 (7.9) 4.676 (7.9)	4.388 (7.9)	4.676 (7.9)	4.302 (7.3)	4.302 (7.3) 4.647 (8.5)	4.266 (6.7) 4.168 (7.9)	4.168 (7.9)
NeuAc,-iso-nLc ₈ $f(9)$ $(2\rightarrow 3)^3$	$(2 \to 3)^3$								

^aFrom the signal of Me₄Si. ^bIn parentheses. For a solution in $di(^2H_3)$ methyl sulfoxide at 308 ±2 K. ^aAgrees with data of Dabrowski et al. ¹¹ except for temperature-dependent shift. ^aAgrees only. ^aAgrees with data of Gasa et al. ¹¹ except for temperature-dependent shift. ^aAgrees only. ^aAgrees with data of Dabrowski et al. ³³ except for temperature-dependent shift.

TABLE II

CHEMICAL SHIFTS (8)4 OF NAC GROUPS AND NEUAC H-3 OF GLYCOSPHINGOLIPIDS^b

Compound	α-NeuA	x-NeuAc- $(2\rightarrow 3)$	a-NeuA	α-NeuAc•(2→6)	α -NeuAc- $(2\rightarrow 3)$	α-NeuAc-(2>6)	GlcNAc-	GlcNAc-	GlcNAc-
The second secon	Н-3е	H-3a	Н-3е Н-3а	Н-За	NAC ₃	NAC ₃	NAC ₃		NAC3
$\operatorname{nL}\!$							1.817		
IV3NeuAcnLc ₄ (2)	2.752	1.367^{c}			1.889			1.817	
$IV^6NeuAcnLc_4(3)$			2.627	1.3144		1.878			1.833
$nLc_6(4)$							1.822 (2)		
$VI^3NeuAcnLc_6(5)$	2.750	1.368			1.888		1.820	1.820	
$VI^6NeuAcnLc_6(6)$			2.624	1.314		1.878	1.817		1.835
iso-nLc ₈ (7)							1.823	1.835	1.843
VI ³ NeuAc-iso-nLc ₈ (8)	2.750	1.366			1.887		1.816	1.834	1.841
VI³, VIII³NeuAc₂-iso-nLc ₈ (9)	2.745	1.372 (2 ea)	a)		1.887(2)		1.815	1.836 (2)	

"From the signal of Me₄Si. For solutions in di(2 H₃)methyl sulfoxide at 308 ± 2 K. $^{c}\alpha$ -Neu Ac.($^{2}\rightarrow 3$) H-4, δ 3.562. $^{d}\alpha$ -Neu Ac.($^{2}\rightarrow 6$) H-4, δ 3.513.

TABLE III

GLYCOSYLATION-INDUCED CHEMICAL SHIFT EFFECTS FOR α -(2 \rightarrow 3) and α -(2 \rightarrow 6) Sialylation of nLc₄ (1), nLc₆ (4), and iso-nLc₈ (7) core glycosphingolipids^a

Group or residue	α-(2→3)	α-(2→6)
β-D-Galp terminal		
H-1	-0.013^{b}	+0.015c
$H-2^d$	-0.010	-0.017
$H-3^d$	+0.573	-0.007
H-4 ^d	+0.065	-0.021
β-D-GlcpNAc subterminal		
H-1e	-0.019	+0.030
$H-2^d$	f	f
$H-3^d$	-0.013	+0.050
NAcf	f	+0.015
β-D-Galp internal		
H-1 ^e	f	f
$H-2^d$	f	f
$H-3^d$	f	+0.032
H-5 ^d	f	f

^aFor a solution in di(²H₃)methyl sulfoxide at 308 K. ^bAverage value for (1→2), (4→5), (7→8), and (8→9). ^cAverage value for (1→3) and (4→6). ^dSingle values for (1→2) or (1→3). ^eAverage values for (1→2), (4→5), or (1→3), (4→6). ^fNot significant [≤0.005].

a pseudotriplet (see Fig. 3) due to virtual coupling between this proton and H-3 of the same sugar ring⁴³⁻⁴⁶, a result of the coincidence of the chemical shifts of H-2 and H-3 (see Table I, compound 3) and the strong coupling between them (compare also the computer-simulated line shapes in Dahmen *et al.*⁴⁷).

Comparison of the data for IV³NeuAcnLc₄ (2) with those of Koerner *et al.*¹² for the ganglioside GM₃, which shares the α -NeuAc-(2 \rightarrow 3)- β -D-Galp-(1 \rightarrow 4) terminal group, suggests two other reporter groups, the subterminal Gal H-3 and H-4 resonances. As illustrated in Fig. 4 for VI³NeuAcnLc₆Cer (5), they are easily found by inspection, which is also true for spectra of similar substances in aqueous media^{41,42}. For these resonances, it does not appear to matter whether the adjacent sugar residue of the aglycon is D-glucose or 2-acetamido-2-deoxy-D-glucose.

There is essential agreement in values for chemical shift changes induced by an α -NeuAc-(2 \rightarrow 3) group on the ring protons of a terminal β -D-galactopyranosyl group given in Table III and those reported for GM₃ by Koerner *et al.*¹², particularly the characteristically large shift for H-3, closest to the glycosylation site, and the somewhat smaller shift change for H-4. The negative shift change for H-2 is not as large as that reported for GM₃. No shift change of such magnitude was found for the same protons upon addition of an α -NeuAc-(2 \rightarrow 6) group, as might be expected. Unfortunately, the effect of this residue on the β -GalIV H-5 and hydroxymethyl protons could not be tested because the appropriate connectivities could not be reliably traced on the COSY spectra available (Fig. 2).

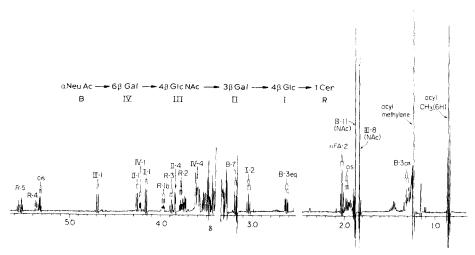


Fig. 3. Downfield region of resolution-enhanced 500-MHz 1 H-n.m.r. spectrum of granulocyte IV 6 NeuAcnLc $_4$ Cer (3) at 308 ± 2 K; conc. ~ 0.5 mM in di(2 H $_3$)methyl sulfoxide-2% D $_2$ O; 936 f.i.d.s. were accumulated.

It is apparent (Table III) that α -(2 \rightarrow 6) sialosylation affects the signal of the nearby GlcNAc NAc ($\Delta\delta$ 0.015 p.p.m.), whereas there is virtually no effect on this resonance from an α -NeuAc-(2 \rightarrow 3)-linked group. The most substantial effect on the signals of the subterminal 2-acetamido-2-deoxy- β -D-glucopyranosyl residue is the shift of H-3 ($\Delta\delta$ 0.50 p.p.m.) upon α -(2 \rightarrow 6) sialosylation. In addition, α -(2 \rightarrow 6) sialosylation has an effect on the *internal* β -D-Gal H-3 that was as large as that on

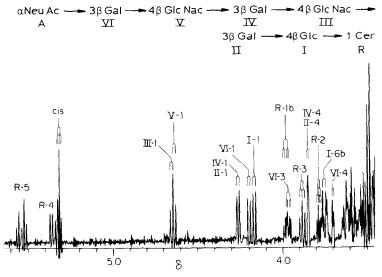


Fig. 4. Downfield region of resolution-enhanced, 500-MHz 1 H-n.m.r. spectrum of granulocyte IV 3 NeuAcnLc $_6$ Cer (5) at 308 ± 2 K; conc. ~ 0.5 mM in di(2 H $_3$)methyl sulfoxide-2% D $_2$ O; 660 f.i.d.s. were accumulated.

the β -D-GlcNAc H-1 ($\Delta\delta$ 0.032 p.p.m.). The effects of α -(2 \rightarrow 3) sialosylation on these protons are all considerably smaller, and, for H-1 and H-3 of β -D-GlcNAc, opposite in sign.

While no attempt was made to elaborate a detailed conformation of the *N*-acetylneuraminyl group from his 13 C-n.m.r. data on sialyl-*lacto*-oligosaccharides, Berman⁴⁸ proposed, on the basis of glycosylation-induced chemical shift effects, that while the α -(2 \rightarrow 3) isomer extends from a lactosyl or lactosaminyl oligosaccharide to form an approximately linear extension of the chain, the α -(2 \rightarrow 6) isomer is folded back over the oligosaccharide in close proximity to the subterminal glucosyl or 2-acetamido-2-deoxyglycosyl residues. Jennings *et al.* ⁴⁹ proposed, based on their 13 C-n.m.r. data for the Type III group B streptococcal polysaccharide, which contains an internal (1 \rightarrow 6)-[α NeuAc-(2 \rightarrow 6)- β -D-Galp-(1 \rightarrow 4)]- β -D-Glcp-NAc-(1 \rightarrow 3)- β -D-Galp structure in its core, an interresidue hydrogen bond between the *N*-acetylneuraminic carboxyl group and the 2-acetamido-2-deoxyglucosyl 3-

Scheme 2. Composite structure of gangliosides derived from *nor*-hexaosylceramide showing approximate disposition of substituents on the nLc_6 core (residues I–VI). The relative conformation of the α -L-Fucp-(1 \rightarrow 3) group (F) was adapted from Hindsgaul *et al.*⁵⁰ for X trisaccharide, and that of the α -NeuAc-(2 \rightarrow 6) group (B) is shown according to the proposal of Jennings *et al.*⁴⁹ for Type III Group B streptococcal polysaccharide.

hydroxy group, stabilizing the conformation of the oligosaccharide determinant. These proposals are illustrated in Scheme 2 of an α -NeuAc-(2 \rightarrow 6) group attached to a poly-N-acetyllactosaminyl chain, e.g., lacto-N-nor-hexaosylceramide. For this discussion, the terminal tetrasaccharide (residues IV, V, VI, and B) is pertinent. For contrast, the approximate position of an α -NeuAc-(2 \rightarrow 3) terminal group (A) is also shown.

The data obtained for compound 3 can be interpreted in terms of this model, although qualifications should be noted regarding (a) the possible differences which may exist in conformation, or in solvation between dimethyl sulfoxide and water solutions of carbohydrates, or both; and (b) the degree to which shift changes <0.2 p.p.m. can be interpreted^{51,52}. Nevertheless, the ¹H-n.m.r. data are consistent with this model in that the protons apparently experiencing the largest shifts to higher

frequency are all on the same side of the oligosaccharide chain, and that the large shift noted for GlcNAc H-3 can be attributed to the close approach of some anisotropic group, which may be the N-acetylneuraminyl carboxyl group, although there is no definite indication for or against the proposed hydrogen bond with HO-3. The shift of internal Gal IV H-3 might be caused by interaction with the N-acetylneuraminyl N-acetyl group. Alternatively, the shifts of Gal IV H-3 and GlcNAc H-3 might be caused indirectly by a change in relative conformation of the GlcNAc N-acetyl group as a result of its interaction with the α -NeuAc-(2 \rightarrow 6) group. However, the lack of a shift change for GlcNAc H-2 is not consistent with this proposal. Interestingly, Jennings et al.49 found that reduction of the sialic acid CO2 to a CH₂OH group caused a shift of the Gal C-1 and GlcNAc C-4 signals of the Nacetyllactosamine residue to which it is linked. This was interpreted as resulting from a conformational change upon elimination of the hydrogen bonding interaction between the carboxyl group and HO-3 of the 2-acetamido-2-deoxyglucosyl residue. However, it is conceivable that the shift changes noted could be attributable solely to the change in chemical environment upon elimination of the anisotropic carboxyl group without invoking a conformational adjustment at those carbon atoms. Berman's conclusions express the same reservations⁴⁸. Although not referring directly to Jennings et al.49 results, Sabesan and Paulson42 also have suggested that the unusual shift changes seen for α -NeuAc-(2 \rightarrow 6) extension of Type 2 oligosaccharides are due to the anisotropic influence of the carboxylate group alone. The calculations performed by Sabesan et al. 53 on α -NeuAc-(2 \rightarrow 6)- β p-Galp- $(1\rightarrow 4)$ - β -p-GlcpNAc- $(1\rightarrow$ show no indication of a hydrogen bond able to control the trisaccharide conformation.

Branched analogs 7-9. — The results in Table I showed that the effect of α -(2 \rightarrow 3) sialosylation on the subterminal Gal H-1 resonances is conserved for the branched mono- and di-N-acetylneuraminyl compounds. In addition, the subterminal Gal H-3,4, NeuAc H-3e,a, and NAc resonances could be readily located (Fig. 5 and Table II). The shifts of the GlcNAc H-1 resonances appeared anomalous. Thus, for compound 8, which had previously! been assigned the structure shown, the most noticeable effects was a shift to higher frequency ($\Delta \delta$ 0.01 p.p.m.) of the β -D-GlcpNAc-(1 \rightarrow 3) H-1, and a somewhat smaller deshielding of the β -D-GlcpNAc-(1 \rightarrow 6) H-1 resonance. Thus, although the existence of a single compound in this fraction could be established from the ¹H-n.m.r. data alone, it would be impossible to assign the NeuAc group to a particular branch, on the basis of a comparison with the spectra of simpler compounds. However, the problem could be apparently solved by comparison with the spectrum of 9, which showed a "normal" shift to lower frequency for the β -D-GlcpNAc-(1 \rightarrow 6) H-1 resonance, thereby suggesting that the precursor 8 had an α -NeuAc-(2 \rightarrow 3)-linked to the $(1\rightarrow 3)$ -branch, and confirming the original structural assignment, which had been made on the basis of analogy to other branched, monosialosylated gangliosides¹. In addition, the β -D-GlcpNAc-(1 \rightarrow 3) H-1 signal showed a further shift to higher frequency ($\Delta \delta$ 0.014 p.p.m.), suggesting that the expected shielding of that proton

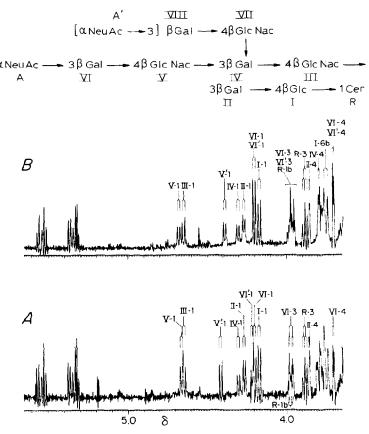


Fig. 5. Downfield region of resolution-enhanced, 500 MHz $^1\text{H-n.m.r.}$, spectra of O erythrocyte: (A) VI³NeuAc-iso-nLc₈Cer (8); and (B), VI³NeuAc-iso-nLc₈Cer (9) at 308 $\pm 2\text{K}$; conc. $\sim\!0.2$ and 0.4mm, respectively, in di(²H₃)methyl sulfoxide–2% D₂O; 2000 and 800 f.i.d.s. were accumulated, respectively. A spectrum identical to that in (A) was obtained with 8 isolated from A erythrocyte membranes.

by the α -NeuAc group (2 \rightarrow 3)-linked to the subterminal Gal residue is being counteracted by some other (deshielding) effect. This may be attributed to interchain interactions resulting from hydrogen bonding of the sialic acid groups, bringing about conformational changes relative to the unsubstituted *iso*-nLc₈ core structure, and locating a deshielding group (*i.e.*, the β -D-GlcpNAc-(1 \rightarrow 6) NAc carbonyl group) near β -D-GlcpNAc-(1 \rightarrow 3) H-1. Evidence for such interchain hydrogen-bonding has been published for the triantennary. N-linked oligo-saccharide structure of fetuin⁵⁴ and assumed to control conformational changes in the branching arms of such structures. Data obtained by Vliegenthart *et al.*⁴¹ also showed that, at least for glycopeptides and related oligosaccharides in D₂O, the effect of α -NeuAc-(2 \rightarrow 3) substitution on a terminal Gal group in several cases causes deshielding rather than shielding effects on H-1 of the Ajacent GlcNAc residue, depending on which branch is substituted. The problem of conformational changes in the system considered here deserves further study.

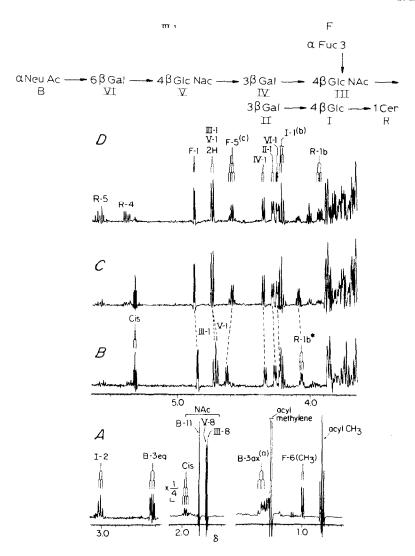


Fig. 6. (A) Upfield region of resolution-enhanced 500-MHz 1 H-n.m.r. spectrum of adenocarcinoma FT75-620 monosialo fraction 6C-upper band (VI6NeuAcIII³FucnLc₆Cer, **10**) at 308 \pm 2K; conc. ~mM in di(2 H₃)methyl sulfoxide-2% D₂O; 400 f.i.d.s. were accumulated. (B) Downfield region of same spectrum. (C) Downfield region at 328 \pm 2K. (D) Downfield region of spectrum of fraction 6C-lower band isolated from same adenocarcinoma, at 328 \pm 2K; conc. ~0.8mM, 900 f.i.d.s. accumulated. (a) NeuAc H-3a confirmed by decoupling from H-3e; (b) Glc H-1 resonances confirmed by decoupling from Glc H-2; (c) Fuc H-5 confirmed by decoupling from Fuc H₃-6 doublet. The unmarked triplet at δ 4.010 p.p.m. in (C) and (D) was assigned to R-3 of sphingosine in combination with 2-hydroxyfatty acid^{12,16,55,57}.

"Fractions 6C". VI⁶NeuAcIII³FucnLc₆Cer (10). — Two monosialosylganglioside fractions, migrating below VI6NeuAcnLc₆Cer (6) on h.p.t.l.c. and reactive with monoclonal antibody IB9 (refs. 5, 6) were isolated as described in the Experimental section. On the basis of the results of methylation analysis, enzymic degradation, and h.p.t.l.c. immunostaining, the structure and chemical VI6NeuAcIII3FucnLc₆Cer (10) was tentatively assigned⁵ to the upper band. ¹H-N.m.r. spectra of the two fractions were obtained, portions of which are displayed in Fig. 6 and summarized in Table IV. It can readily be seen that the spectra of both fractions represent identical carbohydrate components attached to different ceramide aglycons. The following discussion will illustrate how the ¹H-n.m.r. data can be used, with supporting methylation data, to determine the structure of a complex glycolipid without resorting to further degradative procedures. Methylation analysis had previously shown⁵ that Fraction 6C (upper band) contained one 6-O- and two 3-O-linked Gal residues, one terminal Fuc group, 4-O- and 3,4-Olinked GlcNAc residues, and one 4-O-linked Glc residue. Inspection of the ¹Hn.m.r. spectrum revealed one α -anomeric resonance, which could be assigned with confidence to the α -L-Fucp group; and six β resonances, which, taken with the methylation data, suggested a *nor*-hexaosylceramide core structure. The β -D-Glc H-1 resonance was not found in its more typical location around δ 4.16–4.18, but was easily found (by decoupling from the Glc H-2 triplet at δ 3.009) as the doublet at δ 4.213 (${}^{3}J_{1,2}$ 7.3 Hz). At the other end of the saccharide chain, the characteristic NeuAc signals at δ 2.626 (H-3e), 1.308 (H-3a), and 1.876 (NAc) suggested an α -(2→6)-linked terminal NeuAc group, thereby confirming the origin of 2,3,4-tri-Omethyl galactitol (6-O-linked Gal residue) in the g.l.c.-m.s. analysis. This sugar residue was represented in the ¹H-n.m.r. spectrum by the β -D-Galp H-1 resonance (δ 4.227) partially overlapped by the β -D-Glc H-1 resonance at 308 K [compare values for compounds 3 and 6 and lineshape in the ¹H-n.m.r. spectrum of 3 (Fig. 3)]. The H-1 β resonance at δ 4.263 could be assigned to Gal II since this resonance is nearly invariant in the lacto-series ¹H-n.m.r. spectra for a given set of conditions^{9,11,22}. The last H-1 resonance in the β -D-hexose region¹⁸ could be assigned to the β -D-Galp residue in the middle of the chain, with its exact location (δ 4.336) expected to reflect further structural features; the remaining downfield H-1 signals at δ 4.721 and 4.704 (${}^{3}J_{1,2}$ 8.5 Hz) were assigned as β -D-GlcpNAc signals. Again, further structural information can be expected from detailed consideration of their chemical shifts. The presence of two 2-acetamido-2-deoxyglucose residues was further confirmed by a pair of NAc singlets at δ 1.813 and 1.822.

Further specific structural information could be obtained from the 1 H-n.m.r. spectra even in the absence of other data. The chemical shifts of the α -L-Fucp H-1,5, and H₃-6 (δ 4.852, 4.628, and 0.996, respectively, at 308 K; δ 4.878, 4.589 and 1.014, respectively, at 328 K) are fully diagnostic for α -L-Fucp-(1 \rightarrow 3)-D-GlcNAc in Type 2 (X-haptenic) structures^{22,25}. This establishes the origin of the 2-acetamido-2-deoxy-6-O-methylglucitol (3,4-O-linked GlcNAc residue) in the methylation analysis without the necessity of performing the same analysis after

TABLE IV

(10)
, c
<u> </u>
5
Œ
HI3F
E
47
en,
=
-
≎
SS
Ö
0
PRC
_
LED
Ų
3
35
S.
Ξ
9
N 3
(Hz
1) s
INTS (Hz
ANTS (F
ANTS (1
ANTS (F
ANTS (F
NG CONSTANTS (1
NG CONSTANTS (1
NG CONSTANTS (1
ING CONSTANTS (1
COUPLING CONSTANTS (F
NG CONSTANTS (1
)" AND COUPLING CONSTANTS (F
COUPLING CONSTANTS (F
)" AND COUPLING CONSTANTS (F
TS $(\delta)^a$ and coupling constants $(\mathbf{F})^a$
TS $(\delta)^a$ and coupling constants $(\mathbf{F})^a$
$(\delta)^a$ and coupling constants (F
Tal shifts (8)" and coupling constants (F
iical shifts $(\delta)^a$ and coupling constants (F
iical shifts $(\delta)^a$ and coupling constants (F
mical shifts $(\delta)^a$ and coupling constants (F

^aFrom the signal of Me₄Si. ^bIn parentheses. For a solution in di(²H₃)methylsulfoxide at 308 K. ^aAt 338 K. ^bI_{1,2} 7.3 Hz. ^aUpper t.l.c. band major resonances (see Fig. 6B). Other β-D-Glep H-1 resonances found at δ 4.200 and 4.169; H-2 resonances at δ 3.026 and 3.045. The proportions found for the lower t.l.c. band are different. ^fAssignment may be reversed.

defucosylation, as in the previous work⁵, to ascertain whether an α -L-Fucp or β -D-Galp group is linked to O-3 of the β -D-GlcpNAc residue. Furthermore, the precise location of the α -L-Fucp-(1 \rightarrow 3) H₃-6 resonance suggested its attachment to the internal GlcNAc III residue, rather than penultimate GlcNAc V residue, since in the latter case a chemical shift of δ 1.020 \pm 0.001 at 328 K would be observed. Despite the small difference (0.006 p.p.m.), this resonance has been found, in all cases tested, to correlate with the relative position of α -L-Fucp-(1 \rightarrow 3) groups on a poly-N-acetyllactosamine chain²². We realize that the use of such a small effect (3 Hz) understandably invites skepticism, and we would not suggest as a general practice structural assignments based on such a criterion without supporting data. Nevertheless, we believe that the data here and in our previous publication²² show that the difference is not artifactual, and suggests a secondary structural basis²². This assessment of which the D-GlcpNAc residue in the nLc₆ core has the α -L-Fuc-(1 \rightarrow 3) group linked to it was confirmed by considering the alternative where this group is linked to GleNAc V. Since α -(1 \rightarrow 3) fucosylation of a Type 2 chain β -D-GlepNAc residue does not affect the H-1 resonance of the β -D-Galp residue to which the D-GlcpNAc residue is glycosidically linked²², one would in this case expect the two internal β -D-Galp II and IV H-1 resonances to coincide at their original positions relative to nLc₆ (4) at δ 4.26 p.p.m., as they do in Y₂ glycolipid²² (V³FucnLc₆Cer). On the other hand, addition of an α -L-Fucp-(1 \rightarrow 3) group to a β -D-GlcpNAc residue is known to induce the following chemical shift changes (for a solution in dimethyl sulfoxide-2% deuterium oxide at 328 K) for nearby residues in Type 2-chain structures²²: for H-1 of the β -D-GlcpNAc residue to which it is linked, $\Delta \delta 0.060 \pm 0.002$ p.p.m.; for H-1 of the vicinally linked β -p-Galp-(1 \rightarrow 4) residue, $\Delta\delta$ 0.062 \pm 0.005 p.p.m.; and for H-1 of the next β -D-GlcNAc residue linked (1 \rightarrow 3) to that β -D-Galp residue, $\Delta \delta 0.010$ –0.018 p.p.m. For an α -L-fucosyl group (1 \rightarrow 3)-linked to GlcNAc III, addition of these shifts to those given in Table I for compound 6 would give, for H-1 of GlcNAc III, δ 4.716; for that of GlcNAc V, δ 4.701–4.719; and for that of Gal IV, δ 4.326. The values calculated for the two β -D-GlcpNAc H-1 resonances are sufficiently close to the experimental values to permit assignment as shown in Fig. 6 and Table IV. The value calculated for Gal IV H-1 is also reasonably close to the experimental value.

The structure is consistent with all data previously obtained⁵. In addition, the direct-probe, electron-impact mass spectrum of the permethylated compound (not shown) was characterized by an abundant ion pair m/z 825 and 793 representing the terminal sequence NeuAc \rightarrow Hex \rightarrow HexNAc \rightarrow . Finally, the ¹H-n.m.r. spectrum of sialidase-treated 10 was found to be consistent with the expected asialo structure III³FucnLc₆ and distinct from that of V³FucnLc₆²².

Ceramide of Fraction 6C glycolipid (10). — The ¹H-n.m.r. spectra of the two subfractions of 6C revealed some important structural information about the ceramide components of each. In the spectra of the upper band (Figs. 6A, B, and C), the near disappearance of the sphingosine *trans*-vinyl proton signals at δ 5.36 and 5.57, and the concomitant appearance of the major R-1b signal at δ 4.065 (δ

4.083 at 328 K) showed that the long-chain base is predominantly phytosphingosine⁵⁵. Some degree of unsaturation of the fatty acids was indicated by the cis-vinyl proton signal at δ 5.321 and the cis-allyl proton signal at δ 1.977. The absence of any signal between δ 2.0 and 2.2 indicated that the fatty acids have a high proportion of α -hydroxylation, since the α -proton normally resonates in this region¹². A similar ceramide ¹H-n.m.r. spectrum was recorded for GM₂ of mullet roe⁵⁵, although the degree of unsaturation of fatty acids was higher for that glycolipid. Previously, X-reactive (pentaglycosyl)ceramide isolated from adenocarcinoma was found to contain phytosphingosine almost exclusively, along with a high proportion of α -hydroxy fatty acids⁵⁶. The spectra of di- and tri-meric X glycolipids from the same source also showed phytosphingosine signals, and the fatty acid analysis similarly revealed the presence of large proportions of α -hydroxy fatty acids²⁵. This appears to be a rather general characteristic of the majority of adenocarcinoma fucolipids identified so far. Finally, the lower band could be seen (Fig. 6D) to contain a larger proportion of normal sphingosine species (R-1b at δ 3.925 at 328 K, and R-4 and 5 signals of greater intensity), along with a virtual absence of unsaturated fatty acid. The signal at δ 4.010 was assigned to R-3 of unsaturated sphingosine in combination with α -hydroxy fatty acid^{12,16,55,57}. These conclusions were supported by the results of ceramide analysis by g.l.c.-m.s. Thus, estimation of sphingosines as long-chain aldehydes gave, for the upper band, 75% of C₁₈ phytosphingosine (15:0 aldehyde), 25% of C₁₈ sphingosine (16:1 aldehyde), and a trace of C₂₀ phytosphingosine (17:0 aldehyde). For the lower band, the analysis gave 40% of C₁₈ phytosphingosine and 60% of C₁₈ sphingosine. Fatty acid analysis showed the upper band to contain 80% of α -hydroxy fatty acids, almost exclusively as long-chain compounds (22:1, 22:0, 24:1, and 24:0). The lower band yielded 42% of α -hydroxy fatty acids, almost exclusively short-chain compounds (16:0). The variation of the shift position for β -D-Glc H-1 and H-2 (Table IV)) showed that these nuclei are sensitive indicators of the differences in ceramide composition^{16,57}.

Structural basis for accumulation of Fraction 6C glycolipid (10). — The Fraction 6C glycolipid possesses a unique and exemplary structure. Whereas some colonic and liver adenocarcinomata are apparently characterized by induction of abnormal elevations of both α -NeuAc-(2 \rightarrow 6)- and α -L-Fuc-(1 \rightarrow 3)-transferases, the products that one might expect from the fucosylation of GlcNAc V, in addition to or instead of GlcNAc III, have not been found. The terminal tetrasaccharide structure, α -NeuAc-(2 \rightarrow 3)- β -D-Galp-(1 \rightarrow 4)-[α -L-Fucp-(1 \rightarrow 3)]- β -D-GlcpNAc-(1 \rightarrow 3)-, (sialosyl-X) has been found^{5,8} in glycolipids of the same adenocarcinoma as that of Fraction 6C, in glycolipids of kidney⁵⁸, and in a variety of glycopeptide sources⁵⁹⁻⁶¹. Thus, although cases are known where previous attachment of the α -L-Fucp-(1 \rightarrow 3) group will prevent α -(2 \rightarrow 3) sialylation^{62,63}, the sialosyl-X structure can apparently be created by attachment of these sugars in reverse order. By contrast, the isomeric structure, α -NeuAc-(2 \rightarrow 6)- β -D-Galp-(1 \rightarrow 4)-[α -L-Fucp-(1 \rightarrow 3)]- β -D-GlcpNAc-(1 \rightarrow 3), has never been found. Paulson *et al.*⁶⁴ have demonstrated the mutually

exclusive glycosylation by α -NeuAc- $(2\rightarrow 6)$ - and α -L-fucosyl- $(1\rightarrow 3)$ -transferases of asparagine-linked oligosaccharides. In our laboratory, the same phenomenon has been observed with α -L-fuc- $(1\rightarrow 3)$ -transferase (of human lung carcinoma) on α -NeuAc- $(2\rightarrow 6)$ - and α -NeuAc- $(2\rightarrow 3)$ -nLc₆Cer substrates⁶³. Whereas the latter was observed to incorporate two fucosyl groups, the former incorporated only one to produce glycolipid 6C. Glycolipid 6C itself was not further fucosylated under the experimental conditions*.

The stereochemical model supported by n.m.r. data may shed some light on the structural basis for these phenomena. In Scheme 2, the carbohydrate structure of the ganglioside 6C is depicted by residues I–VI and groups B and F. The point of attachment for an incoming α -L-Fucp-(1 \rightarrow 3) group to GlcNAc V is illustrated by an arrow. It can be seen that, even in the absence of the proposed hydrogen-bond, the α -NeuAc-(2 \rightarrow 6) group (B) could represent a barrier to the fucosyl group. The proposed preferred conformation would require the α -NeuAc-(2 \rightarrow 6) and α -L-Fucp-(1 \rightarrow 3) groups to overlap in space in a hypothetical molecule containing both groups. It is apparent, on the other hand, that an α -NeuAc-(2 \rightarrow 3) group (A) would offer no steric hindrance to the fucosylation of GlcNAc V, although it has been observed that previous attachment of the α -L-Fucp-(1 \rightarrow 3) group would prevent α -(2 \rightarrow 3) as well as α -(2 \rightarrow 6) sialosylation⁶²⁻⁶⁴.

In this study, we have attempted to establish a body of ¹H-n.m.r. data useful for the systematic characterization of glycolipids containing α -NeuAc groups (2 \rightarrow 6) as well as $(2\rightarrow 3)$ -linked to the terminal β -D-Galp residue of Type 2-chain lactoseries core structures. Although an extensive compilation of data exists for such structures found in glycopeptides and free oligosaccharides (see for example refs. 41 and 42), the use of a different solvent¹⁸ (dimethyl sulfoxide vs. deuterium oxide) would appear to call for additional data. We have demonstrated, in this series, the occurrence of chemical-shift correlations and systematic trends in glycosidationinduced shift changes which meet the criteria for useful primary structural-reporter groups⁴¹. Because of their regular repeating β -D- $(1\rightarrow 3e)$ - β -D- $(1\rightarrow 4e)$ structure, Type 2-chain oligosaccharides form an extended helix^{66,67} on which terminal and side-chain substituents have effects confined locally (within three residues) and for which, therefore, the application of systematic chemical shift data is particularly useful. This applies also to branched structures, which have been shown to exhibit some specific correlations of their own^{9,15,33}. In addition, in applying 2D n.m.r. methods to the simpler and more abundant analogs available, we have collected additional data consistent with previously proposed stereochemical models^{48,49,53}. These provide not only a working three-dimensional structural hypothesis for the n.m.r. data, but also a basis for rationalizing the accumulation of certain oligosaccharide structures that reflects experimentally established specificities of

^{*}The recent report⁶⁵ of a glycolipid having structure **12** (VI³NeuAcIII³FucnLc₆Cer) isomeric to glycolipid 6C) from chronic myelogenous leukemia cells may imply at least one case of inhibition of α -L-(1 \rightarrow 3)-fucosylation of GlcNAc by previous α -(2 \rightarrow 3)-sialosylation of the β -D-Galp residue (1 \rightarrow 4)-linked to it.

glycosyltransferases as obtained in several systems^{63,64}. It is worth emphasizing that, in general, we agree with the proposition that the shift effects observed in this system are too small to support by themselves a detailed conformational model⁵² (i.e., glycoside torsion-angles specifying the preferred position of the α -(2 \rightarrow 6)linked NeuAc group). Only precise quantitative n.O.e. data or interresidue heteronuclear-coupling constants (or both) could provide the necessary desiderata. None of the n.m.r. data presented so far, here or elsewhere, categorically prove or exclude the existence of a hydrogen bond between the α -NeuAc-(2 \rightarrow 6) carboxyl group and the nearby β -D-GlcpNAc HO-3 which could control the conformation of these sugars. Nevertheless, the grosser aspects of the stereochemistry are supported by ¹³C- and ¹H-n.m.r. data^{42,48,49,53}, HSEA calculations⁵³, and enzyme studies⁶²⁻⁶⁴. In addition, although the argument may continue over the limit at which a change in chemical shift is a useful indicator specifying conformation, the usefulness of a chemical shift difference in specifying primary structure is determined more by its reproducibility than its magnitude. By this criterion, even the difference of 0.006 p.p.m. noted for some α -L-Fucp-(1 \rightarrow 3) H₃-6 resonances can be useful, providing the conclusion is consistent with whatever else is known about the structure. Furthermore, the small magnitude need not necessarily preclude some logical speculation on the source of the effect. This may at least suggest a hypothesis that can be tested by some other method.

Using the data collected for the present study, we have (a) confirmed the structure of two *lacto*-series gangliosides found in polymorphonuclear neutrophils^{3,4} without the need for degradative studies (see Figs. 3 and 4); (b) confirmed details of the structure of the ganglioside G8 previously proposed only on the basis of analogy, but never established by chemical proof¹; and (c) confirmed the structure of an adenocarcinoma fucoganglioside **10** (6C), particularly demonstrating the usefulness of 1D n.m.r. data in drawing conclusions about primary structure without resorting to numerous degradative experiments. In addition, we were able to identify solely on the basis of its ¹H-n.m.r. spectrum a second fraction of 6C not previously characterized and establish approximately the differences in ceramide structure between the two.

Frequently, significant glycolipids from cultured cells, tumors, or normal tissue samples are available only in small quantities ($\sim 100~\mu g$). Thus, it has been necessary to develop methods of structural identification that are rapid and efficient in their use of material, in order to conserve material for other purposes, *i.e.*, production and screening of monoclonal antibodies. Methylation-hydrolysis-g.l.c.-m.s. can now routinely be done on 20–30 μg of material; if ¹H-n.m.r. observations are done first, this procedure could make further degradative procedures, *i.e.*, enzyme treatments, Smith degradation, desialosylation, defucosylation, etc., unnecessary. If an extensive data library is available, ¹H-n.m.r. data alone may be sufficient in certain cases.

ACKNOWLEDGMENTS

The authors thank M. J. Geckle of Bruker Instruments for recording the 300-MHz 2D n.m.r. spectrum, and Eric Holmes for critical reading of the manuscript.

NOTE ADDED IN PROOF

Since the submission of this paper, a structural characterization of several *neolacto*-series gangliosides by methods including 1D ¹H-n.m.r. has been published (T. Taki, K. Matsumoto, K. Yamamoto, T. Matsubara, A. Hayashi, T. Abe, and M. Matsumoto, *Lipids*, 23 (1988) 192–198). In addition, at least two extensive discussions concerning the application of 2D n.m.r. methods to glycoconjugate structure elucidation have become available, which interested readers are advised to consult (T. A. W. Koerner, J. H. Prestegard, and R. K. Yu, *Methods Enzymol.*, 138 (1987) 38–59; J. Dabrowski, *Methods Stereochem. Analysis*, 9 [Two Dimensional N.M.R. Spectroscopy] (1987) 349–386).

REFERENCES

- 1 K. WATANABE, M. E. POWELL, AND S. HAKOMORI, J. Biol. Chem., 254 (1979) 8223–8229.
- 2 O. Nilsson, J.-E. Mansson, E. Tibblin, and L. Svennerholm, FEBS Lett., 133 (1981) 197-200.
- 3 F. W. SYMINGTON, D. L. HEDGES, AND S. HAKOMORI, J. Immunol., 134 (1985) 2498-2506.
- 4 M. N. Fukuda, A. Dell, J. E. Oates, P. Wu, J. C. Klock, and M. Fukuda, J. Biol. Chem., 260 (1985) 1067–1082.
- 5 S. HAKOMORI, E. NUDELMAN, S. B. LEVERY, AND C. M. PATTERSON, Biochem. Biophys. Res. Commun., 113 (1983) 791-798.
- 6 S. HAKOMORI, C. M. PATTERSON, E. NUDELMAN, AND K. SEKIGUCHI, J. Biol. Chem., 258 (1983) 11819–11822.
- 7 IUPAC-IUB COMMISSION ON BIOCHEMICAL NOMENCLATURE, Biochem. J., 171 (1978) 21-35.
- 8 Y. Fukushi, E. Nudelman, S. B. Levery, H. Rauvala, and S. Hakomori, *J. Biol. Chem.*, 259 (1984) 10511–10517.
- 9 J. Dabrowski, P. Hanfland, and H. Egge, Methods Enzymol., 83 (1983) 69-86.
- 10 C. C. SWEELEY AND H. A. NUNEZ, Annu. Rev. Biochem., 54 (1985) 765-801.
- 11 S. GASA, T. MITSUYAMA, AND A. MAKITA, J. Lipid Res., 24 (1983) 174-182.
- 12 T. A. W. Koerner, Jr., J. H. Prestegard, P. C. Demou, and R. K. Yu, *Biochemistry*, 22 (1983) 2676–2687; 2687–2690.
- 13 R. FREEMAN AND H. D. W. HILL, J. Chem. Phys., 54 (1971) 301–313.
- 14 W. P. Aue, E. Bartholdi, and R. R. Ernst, J. Chem. Phys., 64 (1976) 2229-2246.
- 15 J. DABROWSKI AND P. HANFLAND, FEBS Lett., 142 (1982) 138-142.
- 16 A. YAMADA, J. DABROWSKI, P. HANFLAND, AND H. EGGE, Biochim. Biophys. Acta, 681 (1980) 473–479.
- 17 J. H. PRESTEGARD, T. A. W. KOERNER, JR., P. C. DEMOU, AND R. K. YU, J. Am. Chem. Soc., 104 (1982) 4993–4995.
- 18 J. DABROWSKI, P. HANFLAND, AND H. EGGE, Biochemistry, 19 (1980) 5652-5658.
- 19 W. A. GIBBONS, C. F. BEYER, J. DADOK, R. F. SPRECHER, AND H. R. WYSSBROD, Biochemistry, 14 (1975) 420-429.
- 20 R. R. ERNST, Adv. Magn. Reson., 2 (1966) 1-135.
- 21 D. MARION AND K. WÜTHRICH, Biochem. Biophys. Res. Commun., 113 (1983) 967-974.
- 22 S. B. LEVERY, E. D. NUDELMAN, N. H. ANDERSEN, AND S. HAKOMORI, Carbohydr. Res., 151 (1986) 311–328.
- 23 C. C. SWEELEY AND E. A. MOSCATELLI, J. Lipid Res., 1 (1959) 40-47.
- 24 R. C. GAVER AND C. C. SWEELEY, J. Am. Oil Chem. Soc., 42 (1965) 294-298.

 S. Hakomori, E. Nudelman, S. B. Levery, and R. Kannagi, J. Biol. Chem., 259 (1984) 4672– 4680

- 26 B. A. Macher, J. C. Klock, M. N. Fukuda, and M. Fukuda, J. Biol. Chem., 256 (1981) 1968–1974.
- 27 J. R. WHERRET, Biochim. Biophys. Acta, 326 (1973) 63-73.
- 28 B. SIDDIQUI AND S. HAKOMORI, Biochim. Biophys. Acta, 330 (1973) 147-155.
- 29 S. Ando, K. Kon, M. Isobe, and T. Yamakawa, J. Biochem. (Tokyo), 73 (1973) 893-895.
- 30 J. KOŚCIELAK, A. PIASEK, H. GÓRNIAK, A. GARDAS, AND A. GREGOR, *Eur. J. Biochem.*, 37 (1973) 214–225.
- 31 Y. OKADA, R. KANNAGI, S. B. LEVERY, AND S. HAKOMORI, J. Immunol., 133 (1984) 835-842.
- 32 S. K. KUNDU, B. E. SAMUELSSON, I. PASCHER, AND D. M. MARCUS, J. Biol. Chem., 258 (1983) 13857–13866.
- 33 U. Dabrowski, P. Hanfland, H. Egge, S. Kuhn, and J. Dabrowski, J. Biol. Chem., 259 (1984) 7648–7651.
- 34 K. STELLNER, H. SAITO, AND S. HAKOMORI, Arch. Biochem. Biophys., 155 (1973) 464-472.
- 35 M. McNeil and P. Albersheim, Carbohydr. Res., 56 (1977) 239–248.
- 36 R. A. LAINE, Proc. Int. Congr. Pure Appl. Chem., 27th, (1980) 193-198.
- 37 H. BJÖRNDAL, C. G. HELLEROVIST, B. LINDBERG, AND S. SVENSSON, Angew. Chem., Intl. Ed. Eng., 9 (1970) 610–619.
- 38 B. LINDBERG AND J. LÖNNGREN, Methods Enzymol., 50 (1978) 3-33.
- 39 J. F. MAGNANI, D. F. SMITH, AND V. GINSBURG, Anal. Biochem., 109 (1980) 399-402.
- R. KANNAGI, E. NUDELMAN, S. B. LEVERY, AND S. HAKOMORI, J. Biol. Chem., 257 (1982) 14865– 14874.
- 41 J. F. G. VLIEGENTHART, L. DORLAND, AND H. VAN HALBEEK, Adv. Carbohydr. Chem. Biochem., 41 (1983) 209-374.
- 42 S. SABESAN AND J. C. PAULSON, J. Am. Chem. Soc., 108 (1986) 2068–2080.
- 43 J. I. MUSHER AND E. J. COREY, Tetrahedron, 18 (1962) 791-809.
- 44 G. KOTOWYCZ AND R. U. LEMIEUX, Chem. Rev., 73 (1973) 669-698.
- 45 S. J. Perkins, L. N. Johnson, D. C. Phillips. and R. A. Dwek, Carbohydr. Res., 59 (1977) 19–34.
- 46 J. R. BRISSON AND J. P. CARVER, J. Biol. Chem., 257 (1982) 11207-11209.
- 47 J. Dahmen, T. Freid, G. Gronberg, G. Magnusson, and G. Noori, *Carbohydr. Res.*, 125 (1984) 161–164.
- 48 E. BERMAN, Biochemistry, 23 (1984) 3754-3759.
- 49 H. J. JENNINGS, C. LUGOWSKI, AND D. L. KASPER, Biochemistry, 20 (1981) 4511-4518.
- O. Hindsgaul, T. Norberg, J. LePendu, and R. U. Lemieux, *Carbohydr. Res.*, 109 (1982) 109– 142.
- 51 J. Dabrowski, P. Hanfland, H. Egge, and U. Dabrowski, Arch. Biochem. Biophys., 210 (1981) 405–411.
- 52 R. U. LEMIEUX AND K. BOCK, Arch. Biochem. Biophys., 221 (1983) 125-134.
- 53 S. SABESAN, K. BOCK, AND J. C. PAULSON, Proc. Int. Symp. Glycoconjugates VIIth, (1983) 473-474.
- 54 J. C. OBERHOLTZER, S. W. ENGLANDER, AND A. F. HORWITZ, Biochemistry, 20 (1981) 4785–4792.
- 55 Y.-T. LI, Y. HIRABAYASHI, R. DEGASPERI, R. K. YU, T. ARIGA, T. A. W. KOERNER, JR., AND S.-C. LI, J. Biol. Chem., 259 (1984) 8980–8985.
- 56 H.-J. YANG AND S. HAKOMORI, J. Biol. Chem., 246 (1971) 1192–1200.
- 57 J. DABROWSKI, H. EGGE, AND P. HANFLAND, Chem. Phys. Lipids, 26 (1980) 187-196.
- 58 H. RAUVALA, J. Biol. Chem., 251 (1976) 7517-7520.
- 59 G. LAMBLIN, A. BOERSMA, A. KLEIN, P. ROUSSEL, H. VAN HALBEEK, AND J. F. G. VLIEGENTHART, J. Biol. Chem., 259 (1984) 9051–9058.
- 60 T. Krusius and J. Finne, Eur. J. Biochem., 84 (1978) 395-403.
- 61 M. Fukuda, E. Spooncer, J. E. Oates, A. Dell, and J. C. Klock, J. Biol. Chem., 259 (1984) 10925–10935.
- 62 J. Weinstein, U. Desouza-e-Silva, and J. C. Paulson, J. Biol. Chem., 261 (1986) 3737-3743.
- 63 E. H. HOLMES, G. K. OSTRANDER, AND S. HAKOMORI, J. Biol. Chem., 261 (1986) 3737-3743.
- 64 J. C. PAULSON, J.-P. PRIEELS, L. R. GLASGOW, AND R. L. HILL, J. Biol. Chem., 253 (1978) 5617-5624.
- 65 M. N. Fukuda, A. Dell, P. R. Tiller, A. Varki, J. C. Klock, and M. Fukuda, J. Biol. Chem., 261 (1986) 2376–2383.
- 66 E. D. T. ATKINS, D. H. ISAAC, I. A. NIEDUSZYNSKI, C. F. PHELPS, AND J.-K. SHEEHAN, Polymer,